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Batteries
How do battery cells work?

Why?
When several galvanic cells are connected in series, the overall voltage of the battery increases. Batteries are the common solution to this challenge. In a battery, or voltaic cell, individual cells are connected in series, providing a higher potential. In this activity, we will explore the components of a voltaic cell or battery.

Model 1 - Voltaic Cell

1. Consider the reaction in Model 1. Write down the two half-reactions (Ox and Red) for the battery in the left before the wire is connected. Explain why the number of electrons (e⁻) must be the same in both half-reactions.

2. Repeat the same in Model 1, but before the wire is connected and after it is connected. Identify the specific phases of evidence that a chemical reaction has occurred at one point with the wire connected.

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