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Answers to 5118/ 5118 2014 O level Chemistry Paper 1 (MCQ)

Question	Answer	Explanation
21	B	Method 1 \Rightarrow suitable for insoluble gases that will displace water. Method 2 \Rightarrow gases must be less dense than air so that it does not escape from the gas jar
22	D	Zinc ion forms white ppt with aqueous ammonia & ppt dissolves in excess. A sulfate forms a white ppt when barium nitrate is added. Lead ion forms white ppt with aqueous ammonia & ppt does not dissolve in excess.
23	A	Known fact
24	A	Number of neutrons = $85 - 37 = 48$ Rb^+ \Rightarrow atom has lost one electron so $37 - 1 = 36$
25	C	Calcium fluoride is an ionic compound (metal + non-metal). Calcium \Rightarrow 2.8.8.2 & Fluorine \Rightarrow 2.7
26	D	Known fact
27	D	$2Na + Cl_2 \rightarrow 2NaCl$
		20 cm ³ of ethene will react with 60 cm ³ of oxygen producing 40

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